

MANICALAND STATE UNIVERSITY OF APPLIED SCIENCES

FACULTY OF ENGINEERING

DEPARTMENT: CHEMICAL AND PROCESSING ENGINEERING

MODULE: TRANSPORT PHENOMENA

CODE: CHEP 211

SESSIONAL EXAMINATIONS DECEMBER 2022

DURATION: 3 HOURS

EXAMINER: MISS N. T. MADZIWA

INSTRUCTIONS

- 1. Answer all questions.
- 2. Start a new question on a fresh page
- 3. Total marks 100
- 4. Formulae sheet is given at the end of the paper.

Additional material(s): Calculator and Lennard-Jones
Constants Tables

QUESTION 1

- a) State Fick's law of molecular diffusion and define each term. [3]
- b) Show that for diffusion in gases, the diffusivities of species **A** and **B**, D_{AB} and D_{BA} are equal. [6]
- c) Describe how heat transfer by conduction occurs. [4]
- d) Give *two* differences between forced convection and free convection in heat transfer. [2]
- e) Under what conditions does the Navier-Stokes equation reduce to:
 - i. Stoke's or Creeping flow. [1]
 - ii. Euler equation. [1]
- f) By performing a mass balance on a volume element $(\Delta x)(\Delta y)(\Delta z)$, show that for a fluid of constant density, the equation of continuity reduces to:

$$\frac{\partial v_x}{\partial x} + \frac{\partial v_y}{\partial y} + \frac{\partial v_z}{\partial z} = 0 \text{ or } \nabla \cdot v = 0$$
 [8]

QUESTION 2

A thin film of liquid of density ρ and viscosity μ flows down a vertical wall of length L and width W. The film thickness is δ . By performing a shell momentum balance on the z-component of momentum,

- a. derive an equation for the velocity profile of the film. [20]
- b. hence, determine the maximum velocity. [5]

QUESTION 3

- a) An uninsulated steel pipe in a building transports steam at 450 °C. The walls of the building and the surrounding air are both at 25 °C. State and explain the heat transfer mechanisms that occur in this system. [6]
- b) A furnace wall consists of 30 cm of fire brick (k = 1 W/m·K), 25 cm of insulating brick (k = 0.12 W/m·K) and a 35 cm of red brick (k = 0.75 W/m·K). The inner wall of fire brick is exposed to furnace gas at 1200K whilst air at 310 K is adjacent to the outside wall of red brick. The inside and outside convective heat transfer coefficients are 95 and 20 W/m²·K, respectively. Determine:
 - i. the heat loss per square metre of the composite wall. [7]
 - ii. the temperature of the innermost wall surface. [2]
- c) Calculate the rate of cooling of an aluminium plate (1.5 m by 1.5 m) at 580 K when it is suspended horizontally in stagnant air at 300 K. The heat transfer coefficient is estimated using the following correlation:

$$Nu_L = 0.14 \ Gr_L^{\frac{1}{3}} Pr^{\frac{1}{3}}$$

where
$$Gr_L = \frac{L^3 \rho^2 g \beta \Delta T}{\mu^2}$$

The properties of air, evaluated at the film temperature of 440 K are:

$$\rho = 0.8021 \text{ kg/m}^3 ; c_p = 1.0197 \times 10^3 \text{ J/kg·K} ; \mu = 2.4453 \times 10^{-5} \text{ Pa·s} ;$$

$$k = 3.6427 \times 10^{-2} \text{ W/m·K} ; \beta = 2.2681 \times 10^{-3} \text{ K}^{-1}$$
[10]

QUESTION 4

- a) Outline the mass transfer of a component between two immiscible phases using the Two-Resistance Theory, clearly stating the assumptions. [5]
- b) Using the Two-Resistance Theory, show that for steady state mass transfer from a gas phase to a dilute liquid phase, the overall mass transfer coefficient

in the liquid phase, K_L is related to the individual mass transfer coefficients of the gas phase and liquid phase, k_G and k_L respectively, by the following expression:

$$\frac{1}{K_L} = \frac{1}{k_L} + \frac{1}{Hk_G}$$

Where *H* is the Henry's Law constant

[7]

c) In the absorption of component A from an air stream into an aqueous stream, the composition of the two adjacent streams were $C_{AL} = 4 \text{ kgmol/m}^3$ and $P_{AG} = 1.013 \times 10^4 \text{ Pa}$. The Henry's law constant for this system is $1.674 \times 10^3 \text{ Pa/(kgmol/m}^3)$. The overall liquid phase mass transfer coefficient, K_L was equal to $1.26 \times 10^{-6} \text{ kgmol/m}^2 \cdot \text{s} \cdot (\text{kgmol/m}^3)$. If 60% of the total resistance to mass transfer is encountered in the liquid film, determine the individual film mass transfer coefficients, K_G and K_L , the molar flux of K_L and the interfacial concentration K_L .

END OF EXAMINATION

FORMULAE SHEET

Diffusional molar flux in the z-direction: $N_{Az} = -cD_{AB} \frac{\partial y_A}{\partial z} + y_A N_{total}$

Diffusional molar flux of A in the z-direction: $N_{A_z} = -CD_{AB} \frac{\partial y_A}{\partial z} + y_A (N_{A_z} + N_{B_z})$

Navier-Stokes equation: $\rho \frac{Dv}{Dt} = -\nabla P + \mu \nabla^2 v + \rho g$

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Navier-Stokes equation in the *x*-direction:

$$-\frac{\partial P}{\partial x} + \mu \left(\frac{\partial^2 v_x}{\partial x^2} + \frac{\partial^2 v_x}{\partial y^2} + \frac{\partial^2 v_x}{\partial z^2} \right) + \rho g_x = \rho \left[\frac{\partial v_x}{\partial t} + v_x \frac{\partial v_x}{\partial x} + v_y \frac{\partial v_x}{\partial y} + v_z \frac{\partial v_x}{\partial z} \right]$$

Equation of Continuity:
$$\frac{\partial v_x}{\partial x} + \frac{\partial v_y}{\partial y} + \frac{\partial v_z}{\partial z} = 0$$

Stoke's flow or Creeping flow: $-\nabla \mathbf{P} + \mu \nabla^2 \mathbf{v} + \rho \mathbf{g} = 0$

Euler equation: $\rho \frac{Dv}{Dt} = -\nabla P + \rho g$

Grad function: $\nabla = \frac{\partial}{\partial x}i + \frac{\partial}{\partial y}j + \frac{\partial}{\partial z}k$

Velocity in vector notation: $\mathbf{v} = v_x i + v_y j + v_z k$

Gravitational acceleration in vector notation: $\mathbf{g} = g_x i + g_y j + g_z k$

z-component of momentum flux in x-direction: $\phi_{xz} = \tau_{xz} + \rho v_x v_z$

z-component of momentum flux in y-direction: $\phi_{yz} = \tau_{yz} + \rho v_y v_z$

z-component of momentum flux in z-direction: $\phi_{zz} = P + \tau_{zz} + \rho v_z v_z$

General formula for viscous stress tensors: $\tau_{ij} = -\mu \left(\frac{\partial v_i}{\partial j} + \frac{\partial v_j}{\partial i} \right)$ where i, j = x, y, z