

MANICALAND STATE UNIVERSITY OF APPLIED SCIENCES

FACULTY OF AGRIBUSINESS AND COMMERCE

DEPARTMENT OF AGRICULTURE ECONOMICS

AND

DEVELOPMENT

INTRODUCTION TO SOIL SCIENCE CODE: AEDT125

SESSIONAL EXAMINATIONS OCTOBER 2021

DURATION: 3 HOURS

EXAMINER: Mr N. SAKADZO

INSTRUCTIONS

- 1. Answer any **Four** questions
- 2. Be concise and clear
- 3. Total marks 100

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-	ion 1	
• .	Discuss the following factors that influence soil formation:	
i)	Organisms	[5]
ii)	Time	[5]
	Describe properties of a named 1:1 aluminosilicate clay	[10]
c)	With aid of a diagram describe the structure of montmorillonite	. [5]
Ques	ion two	
a)	Define soil pH.	[2]
b)	Explain the notion that aluminium ions act as a source of indire ions (H^+) and therefore become a source of acidity in soils.	ct hydrogen [8]
c)	Indicate, giving reasons the pH values that would you expect for following soils:	or the
	i) Soils in arid areas receiving less than 500mm rainfall per	annum. [5]
	ii) Soils in humid climate receiving about 1200mm rainfall	per annum [5]
d)	Differentiate between a pH dependent charge and a permanent aluminosilicate clays.	charge in [5]
Ques	ion three	
	a) A farmer intended to grow maize on his farm. He collected a using a core measuring 100mm in diameter and 60mm heigh drying, the mass of the soil sample was 520g.	1
	i) Calculate the bulk density of the soil in Mg m^{-3} .	[7]
	ii) Explain the low bulk densities exhibited in clay soils.	[2]
	b) Discuss two factors which affect soil aeration.	[10]

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c) Describe three ways which destroy soil structure. [6]			
Question four			
a) Explain why nitrogen fertilizer is susceptible to leaching in soils. [4]			
$1) D_{1} = u^{2} + \frac{1}{2} + 1$			
b) Describe three methods that can be used to control nitrogen fertilizer loss by leaching in cropping systems. [6]			
c) Explain five strategies that can be used to enhance phosphorous availability to			
crops in light sandy soils.			
[15]			
Question five a) Given			
Exchangeable Ca (mmoles _c Kg ⁻¹) =50			
Exchangeable Mg $=45$			
Exchangeable K =3			
Exchangeable Na =1			
Exchangeable H =12			
Exchangeable Al =39			
Calculate the following:			
i) Cation Exchange Capacity (CEC) [2]			
ii) Total Exchangeable bases (TEB) [2]			
iii) % Base Saturation [2]			
iv) Sodium Adsorption Ratio (SAR) [2]			
v) Exchangeable Sodium Percentage (ESP) [2]			
b) Describe the two steps that are involved in the process of nitrification in soils.			
[8]			
c) Write short notes on the following:			
i) Symbiotic Biological Nitrogen Fixation [4]			
ii) Mycorrhizae Symbiosis [3]			
END OF EXAMINATION			
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